## organic papers

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#### **Key indicators**

Single-crystal X-ray study T = 113 K Mean  $\sigma$ (C–C) = 0.003 Å R factor = 0.041 wR factor = 0.104 Data-to-parameter ratio = 9.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# 5-*tert*-Butyl-3-{[6-(2,2-dimethyl-1,3-dioxolan-4-yl)-2,2-dimethyltetrahydrofuro[3,4-*d*][1,3]dioxol-4-yloxy]methyl}-2-hydroxybenzaldehyde

Theare are two independent molecules in the asymmetric unit of the title compound,  $C_{24}H_{34}O_{8}$ . The crystal packing is mainly determined by weak intermolecular  $C-H\cdots O$  hydrogen bonds and van der Waals interactions. Received 12 June 2006 Accepted 11 July 2006

## Comment

Chiral manganese-salen complexes [salen is bis(salicylidene)ethylenediamine] are effective catalysts owing to their ability for highly stereoselective epoxidation of unfunctionalized alkenes (Lane & Burgess, 2003). In order to test the effects of carbohydrates on the asymmetric epoxidation of olefins, we have designed and synthesized a series of chiral salicyaldehyde derivatives, ligands and complexes. The structure of 3-tert-butyl-5-{[5-(2,2-dimethyl-1,3-dioxolan-4-yl)-2,2dimethyltetrahydrofuro[2,3-d][1,3]dioxol-6-yloxy]methyl}-2hydroxybenzaldehyde has been reported previously (Zhao et al., 2006). In this paper, we present the results of our study of another derivative of salicyaldehyde, (I), obtained by the condensation of 5-tert-butyl-3-(chloromethyl)-2-hydroxybenzaldehyde and 2,3:5,6-di-O-isopropylidede-α-D-mannofuranose.



The asymmetric unit of (I) contains two independent molecules (Fig. 1), which show normal bond distnaces and angles (Allen *et al.*, 1987). The distances between the newly bonded atoms C13–O6 and C37–O14 are 1.442 (3) and 1.433 (3) Å, respectively. In the crystal structure, weak intermolecular C–  $H \cdots O$  hydrogen bonds (Table 1) contribute to the stabilization of the crystal packing (Fig. 2), along with van der Waals interactions.

## **Experimental**

The title compound was prepared according to the procedure of Zhao *et al.* (2006). Colourless single crystals of (I) were grown by slow evaporation of a petroleum ether–EtOAc (5:1 v/v) solution. The structure of (I) (m.p. 353–354 K) was confirmed by optical rotation { $[\alpha]_D^{20} = +85.0^{\circ}$  (EtOH)}, elemental analysis, NMR and IR data.

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#### Crystal data

 $C_{24}H_{34}O_8$   $M_r = 450.51$ Monoclinic,  $P_{2_1}$  a = 12.7880 (19) Å b = 6.7749 (10) Å c = 26.859 (4) Å  $\beta = 92.118 (2)^{\circ}$   $V = 2325.4 (6) \text{ Å}^3$ 

## Data collection

Rigaku Saturn CCD area-detector diffractometer  $\omega$  scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)  $T_{\min} = 0.970, T_{\max} = 0.985$ 

#### Refinement

Refinement on  $F^2$   $R[F^2 > 2\sigma(F^2)] = 0.041$   $wR(F^2) = 0.104$  S = 1.005939 reflections 598 parameters

#### Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdots A$	$D \cdots A$	$D-\mathrm{H}\cdots A$
O7-H7···O8	0.84 (3)	1.84 (3)	2.618 (2)	152 (3)
$C13-H13B\cdots O4^{i}$	0.99	2.35	3.330 (3)	172
$C37-H37B\cdots O12^{i}$	0.99	2.44	3.398 (3)	163

Z = 4

 $D_x = 1.287 \text{ Mg m}^{-3}$ 

Mo  $K\alpha$  radiation

Block, colourless

 $0.32 \times 0.20 \times 0.16 \text{ mm}$ 

20584 measured reflections

5939 independent reflections

4809 reflections with  $I > 2\sigma(I)$ 

H atoms treated by a mixture of

 $w = 1/[\sigma^2(F_o^2) + (0.0633P)^2]$ where  $P = (F_o^2 + 2F_c^2)/3$ 

independent and constrained

 $\mu = 0.10 \text{ mm}^{-1}$ 

T = 113 (2) K

 $R_{\rm int} = 0.038$ 

 $\theta_{\rm max} = 27.9^\circ$ 

refinement

 $(\Delta/\sigma)_{\rm max} = 0.001$ 

 $\Delta \rho_{\text{max}} = 0.36 \text{ e } \text{\AA}^{-3}$  $\Delta \rho_{\text{min}} = -0.23 \text{ e } \text{\AA}^{-3}$ 

Symmetry code: (i) x, y - 1, z.

The H atom of the hydroxy group was located in a difference Fourier map and refined, with an O–H distance restraint of 0.84 (3) Å. All other H atoms were included in calculated positions, with C–H = 0.95–1.00 Å and  $U_{\rm iso}(\rm H) = 1.2U_{\rm eq}(\rm C)$  or  $1.5U_{\rm eq}(\rm methyl$ C). In the absence of significant anomalous scattering, Friedel pairs were merged prior to the final refinement; the absolute configuration is known from that of the mannofuranose reagent.

Data collection: *CrystalClear* (Rigaku/MSC, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *CrystalStructure* (Rigaku/MSC, 2005); software used to prepare material for publication: *CrystalStructure*.

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#### Figure 1

The asymmetric unit of (I), with displacement ellipsoids drawn at the 30% probability level. H atoms are shown as small spheres of arbitrary radii.



#### Figure 2

A packing diagram of (I), viewed along the a axis. Dashed lines denote intermolecular hydrogen bonds

## References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1–19.
- Lane, B. S. & Burgess, K. (2003). Chem. Rev. 103, 2457-2473.
- Rigaku/MSC (2005). CrystalStructure (Version 3.7.0) and CrystalClear (Version 1.3.6). Rigaku/MCS, The Woodlands, Texas, USA.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Zhao, S.-Ś., Zhao, J.-Q. & Zhao, D.-M. (2006). Acta Cryst. E62, o2537-o2538.